

Goals for Today's Lecture

study multiplicity for big Einstein solid containing
two macroscopic interacting Einstein solids

introduce fundamental assumption of statistical physics:
all accessible microstates of isolated system are equally likely

show how fundamental assumption leads to several conclusions:

- SZ spontaneously evolves to maximum # of microstates
- heat transfer is irreversible
- heat transfer is not a law of nature but a consequence of probabilities

how to handle large factorials by Stirling's formula

show that width of multiplicity is $1/\sqrt{N}$ so exceedingly
narrow for $N \sim 10^{23}$

Key discoveries of last lecture

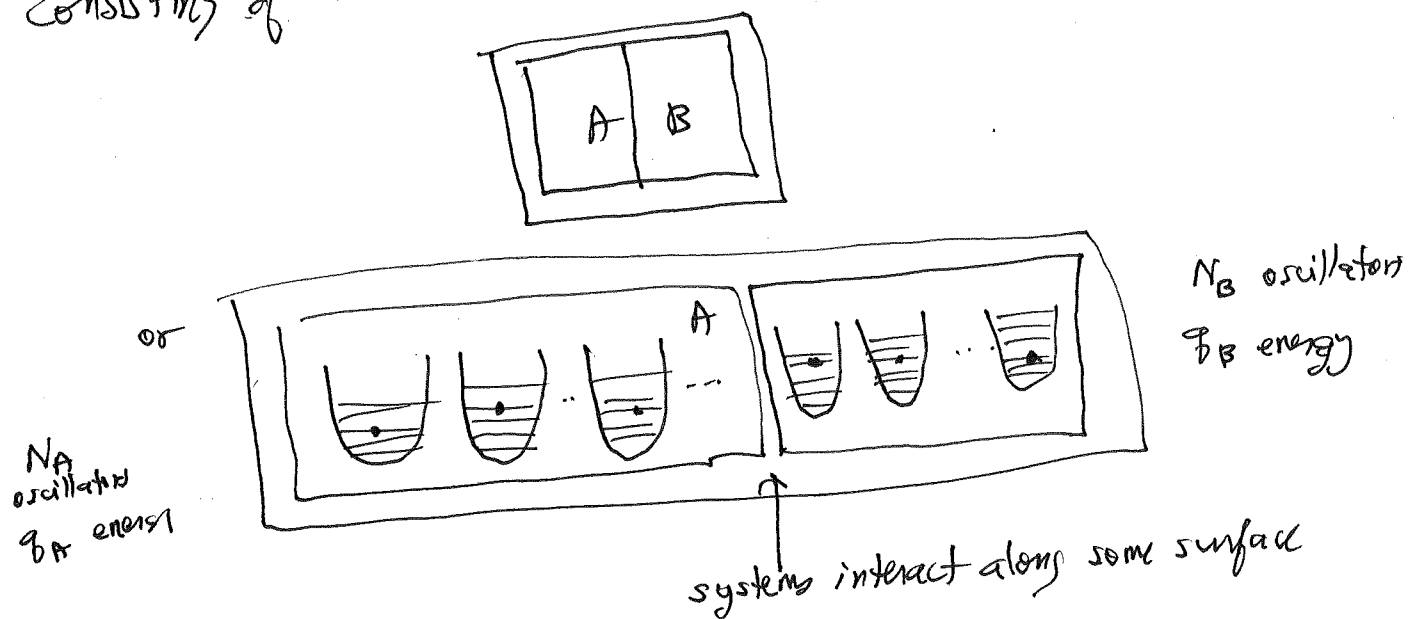
$$\Omega(N, q) = \binom{N-1+q}{q}$$

N = # of identical independent quantum harmonic oscillators
 q = total energy of all N oscillators, $q = q_1 + \dots + q_N$

Einstein Solid With Two Interacting Subsystems

(2)

To understand why entropy evolves to maximum, why there is "spontaneous" irreversible transfer of heat from high temp to low temp subsystems, look at isolated Einstein solid consisting of two macroscopic interacting subsystems



$$\# \text{ of atoms in surface} \propto L^2$$

$$\# \text{ of atoms in volume} \propto L^3$$

$$\frac{\text{energy of interaction}}{\text{energy of subsystem}} \sim \frac{A}{V} = \frac{1}{L}$$

so for local interactions, just between nearby atoms and oscillators, interaction is weak, transfer of energy is slow.

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For weakly interacting subsystems, energy of each subsystem changes slowly over time, of order L^2/k where k is thermal diffusivity ($\sim 10^{-5} \text{ m}^2/\text{s}$ for many solids).

So over short time intervals $< L^2/k$, can consider energies q_A and q_B of two subsystems constant

so can define macrostate of two subsystems by value of q_A since q_B known by energy conservation

$$q_A + q_B = q = \text{total energy}$$

q_A, q_B, q non-negative integers

multiplicity of A: $\Omega_A = \Omega(N_A, q_A) = \binom{N_A - 1 + q_A}{q_A}$

multiplicity of B: $\Omega_B = \Omega(N_B, q_B) = \binom{N_A - 1 + q - q_A}{q - q_A}$ $\sin u$
 $q_B = q - q_A$

multiplicity of entire isolated solid.

$\Omega_{\text{total}} \approx \Omega_A \times \Omega_B$

since, for weakly interacting subsystems, A and B are statistically independent so # microstates in A independent of # microstates in B

Ω is multiplicative over subsystems!

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Discusses Fig. 2.5 on page 59 of Schroeder

Show Mathematica code coupled-einstein-solids.nb
on projector

assume $N_A = 300$ oscillators

$N_B = 200$ oscillators

$q = 100$ energy units (actual $U = h\nu q$)
harmonic oscillator
frequency

total number of macrostates is 101

$q_A = 0, 1, \dots, 99, 100$

total number of microstates is $\binom{N_A + N_B - 1 + q}{q}$

} pool all
oscillators
together to
get this #

$$= \binom{499 + 100}{100} \sim 10^{116}$$

an enormous number compared to # of macrostates

IF we put A together with B and wait, we expect uniform
distribution of energy so expect equilibrium solid to have

$$q_A \text{ in equilibrium} = \frac{300}{300 + 200} \times 100 = 60$$

$$q_B \text{ in equilibrium} = \frac{200}{300 + 200} \times 100 = 40$$

Work out multiplicity $\Omega_{\text{total}}(E_A)$

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q_A	$\Omega_A = \binom{N_A - 1 + q_A}{q_A}$	q_B	$\Omega_B = \binom{N_B - 1 + q - q_B}{q - q_B}$	$\Omega_{\text{total}} = \Omega_A \Omega_B$
0	1	100	$2.8 \cdot 10^{81}$	$3 \cdot 10^{81}$
1	300	99	$9.3 \cdot 10^{81}$	$2.8 \cdot 10^{83}$
2	45/50	98	$3.1 \cdot 10^{80}$	$1.4 \cdot 10^{85}$
⋮		⋮		
100	$1.7 \cdot 10^{96}$	0	1	$1.7 \cdot 10^{96}$

sum of Ω_{total} column = $\binom{N_A + N_B - 1 + q}{q} \approx 10^{116}$

see Schroeder p. 59 and Mathematica code

$$\max(\Omega_{\text{total}}) = \Omega_{\text{total}}(60) \approx 6.9 \cdot 10^{114}$$

$$\min(\Omega_{\text{total}}) = \Omega_{\text{total}}(0) \approx 2.8 \cdot 10^{81} \leftarrow \text{huge \# even for smallest multiplicity}$$

$$\frac{\max(\Omega_{\text{total}})}{\min(\Omega_{\text{total}})} \approx 10^{33}$$

some macrostates enormously more likely than other macrostates because # of accessible microstates differ by so much

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Connection of multiplicities to physics made
by fundamental assumption of statistical physics

"all accessible microstates of isolated system are equally likely"

But tricky how to interpret words "equally likely"

(1) if you take a snapshot of system at any time, each microstate equally likely to be observed at that time requires sense of rapid random evolution, in which macroscopic system fluctuates or flickers or transitions endlessly from one consistent microstate to another consistent microstate

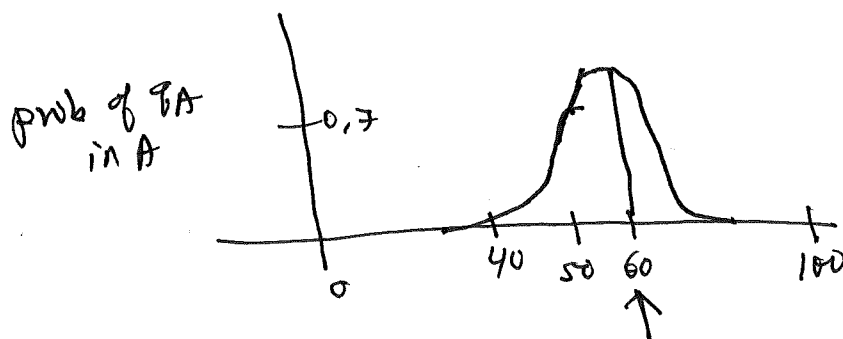
(2) Or consider an ensemble of K identical isolated Einstein solids, each with $N_A + N_B$ oscillators, each with total energy q . Then we can look at all K solids at one time and see many different accessible microstates

Fundamental assumption allows us to convert multiplicities $\Omega_{\text{total}}(q_A)$ into probability of observing Einstein solid with energy q_A in A, energy $q - q_A$ in B

$$\text{prob}(q_A \text{ in A}) = \frac{\Omega_{\text{total}}(q_A)}{\Omega_{\text{total}}} \quad \left. \vphantom{\frac{\Omega_{\text{total}}(q_A)}{\Omega_{\text{total}}}} \right\} \begin{array}{l} \text{total \# of} \\ \text{microstates for} \\ A+B \end{array}$$

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Look at Mathematica code
 coupled-einstein-solids.nb
 and calculation of probabilities. Get plot



prob hits peak at $q_A = 60$, exactly as we expect based on uniform distribution of energy

prob decreases rapidly away from $q_A = 60$

$$\text{prob}(q_A < 30), \text{prob}(q_A > 90) \sim 10^{-6}$$

$$\text{prob}(q_A < 10) \sim 10^{-20}$$

↑ unobservable over age of universe

$$10^9 \text{ y} \times 10^{17} \frac{\text{s}}{\text{y}} \approx 10^{26}$$

Conclude: if all microstates of Einstein solid equally likely, only energy partitions with q_A close to 60 likely

- (1) system evolves spontaneously toward max of Σ_{total}
- (2) transfer of heat is irreversible, very probable

Have discovered entropy: $S = k \ln \Omega$ (8)

Ω is multiplicative over weakly interacting subsystems

$$\Omega_{\text{total}} \approx \Omega_A \times \Omega_B$$

$$\begin{aligned} \text{so } S_{\text{total}} &= k \ln [\Omega_A \Omega_B] = k \ln \Omega_A + k \ln \Omega_B \\ &= S_A + S_B \end{aligned}$$

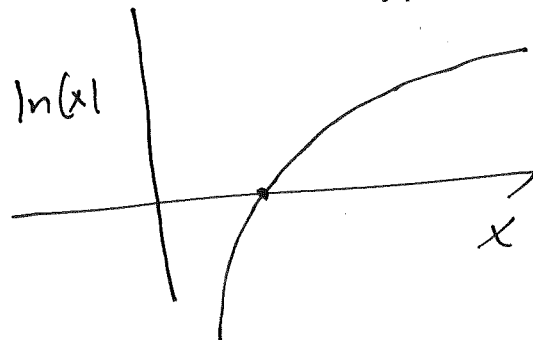
entropy is additive since Ω multiplicative

Further: $\frac{dS}{dt} > 0$ for nonequilibrium isolated systems

reason is that Ω steadily increases towards Ω_{max} as microstates with most accessible microstates are much more likely

$S = k \ln \Omega$ also increases to maximum since $\ln(x)$

is monotone increasing: $\frac{d \ln(x)}{dx} = \frac{1}{x} > 0$ for $x > 0$



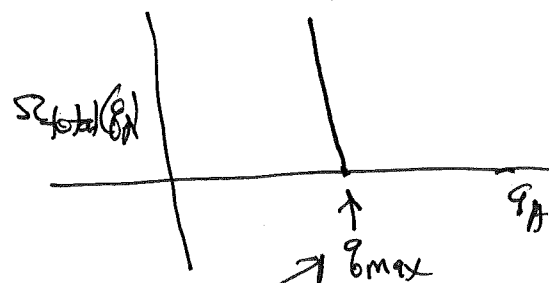
slope everywhere positive

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Next Big Insight:Width of $S_{\text{total}}(q_A) \propto \frac{1}{\sqrt{N}}$ where $N = \text{total \# of oscillators}$

In English: as # of oscillators approaches big # of order $N_A \sim 10^{23}$, multiplicity of Einstein solid becomes fantastically narrow, $1/\sqrt{N} \sim 10^{-10}$

essentially never see any arrangement of energy between two subsystems except for uniform distribution



what curve looks like for $N_A \sim N_B \sim 10^{23}$

$$q_A = \left(\frac{N_A}{N_A + N_B} \right) q = q_{\text{max}}$$

$$q_B = \left(\frac{N_B}{N_A + N_B} \right) q$$

can only observe one state (to 10 digits) with maximum number of microstates.

To derive this result, need to learn how to approximate binomial coefficients $\binom{N}{k}$ when $N \sim 10^{23}$, $k \sim 10^{23}$
surprisingly, this is not so hard to do

Working With Big Numbers

To understand Einstein solids for $N \sim 10^{23}$ need
to find ways to approximate $S(N, q) = \binom{N+q-1}{q}$
for $N, q \gg 1$

Useful to think about different sizes of #s:

small or human #s:

1-1,000

can count comfortably
money you would notice
missing

large #s: $l = 10^5$

↓
small #
↑
large #

Astro 10²³

stars in galaxy 10¹¹ = # neurons in human brain
 googol 10¹⁰⁰ = # galaxies in universe

note: 70! $\approx 10^{100}$

moves in chess game $\sim 10^{120}$

10²⁶ years time for proton to decay

machine precision $\frac{1}{10^{16}}$

large #s have unusual property that

$$\boxed{s + l = l}$$

e.g. $10 + 10^{23} \approx 10^{23}$

small number added to
large # same as large #

$10^{-17} + 1 = 1$ on computer

implies that one should add
power series backwards,
smallest term first

$$\underline{x^{100} + x^{99} + \dots + x + 1}$$

Very Large Numbers

$$V = 10^1$$

examples $52 \sim 10^{23}$

googolplex $10^{10^{100}}$

Very large numbers have property

$$L \cdot V = V$$

large $\times$ very large = very large

E.g. $10^{23} \cdot 10^{10^{100}} = 10^{23 + 10^{100}} = 10^{10^{100}}$

Biggest number used in serious way in science or mathematics is Graham's #, extremely difficult to grasp

hint to understand $\mathcal{G}_1$, first of 64 recursions of 'exponentiation' algorithm:

$$\text{tower } 1 = 3$$

$$\text{tower } 2 = 3^{(3^3)} \quad \text{height of tower given by tower 1}$$

$$\text{tower } 3 = 3^{3^{3^{\dots 3}}} \quad \leftarrow \# \text{ of } 3\text{'s is } 3^{3^3}$$

$$\text{tower } 4 = 3^{3^3 \dots 3} \quad \# \text{ of } 3\text{'s given by tower 3}$$

$\vdots$

$$\text{tower } n = 3^{3^{3^{\dots 3}}} \quad \# \text{ of } 3\text{'s given by tower } n-1$$

where n = value of 3rd tower

$\mathcal{G}_1$ is starting point for constructing Graham's #

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Stirling's Formula

Back to problem of how to estimate $\binom{N}{k} = \frac{N!}{k!(N-k)!}$

for N, k large numbers, 10^5 or, say 10^{23}

Use gift of 17th and 18th century mathematics from de Moivre and Stirling called Stirling's formula

$$n! = 1 \cdot 2 \cdot \dots \cdot n \approx \sqrt{2\pi n} \left(\frac{n}{e}\right)^n = \text{St}(n)$$

already accurate for $n \geq 5$

see Mathematica plot of relative error

$$\frac{n! - \text{St}(n)}{n!} \text{ vs } n$$

Note: for n large, $\left(\frac{n}{e}\right)^n$ is a very large #
 $\sqrt{2\pi n}$ is just a large #

so $n! \approx \left(\frac{n}{e}\right)^n$ for $N \sim 10^{23}$

$$\Rightarrow \boxed{\ln(n!) = n \ln n - n}$$

widely used form
of Stirling's formula

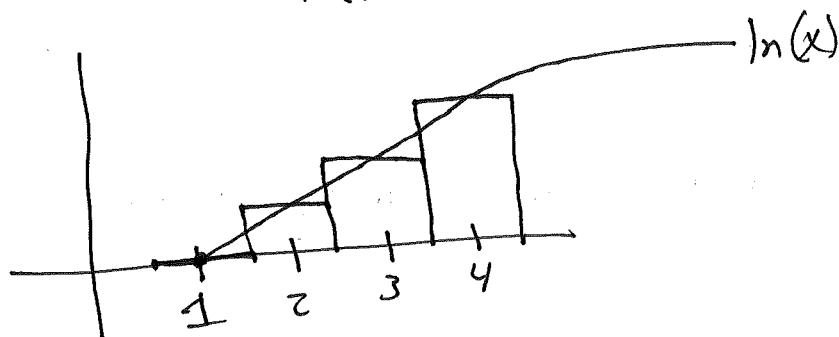
$$n! \approx e^{n \ln n - n} = e^{n(\ln n - 1)}$$

grows a bit faster than
exponentially with n

$$70! \approx 10^{100} = 1 \text{ googol}$$

Origin of Stirling's Formula

Look at $\ln(n!) = \ln(1 \cdot 2 \cdot 3 \cdots n)$
 $= \ln(1) + \ln(2) + \cdots + \ln(n)$



can guess that $\ln(1) + \ln(2) + \ln(3) + \cdots$ is roughly area under $\ln(x)$ from 1 to n

$$\ln(n!) = \sum_{i=1}^n \ln(i) \approx \int_1^n \ln(x) dx = x \ln(x) - x \Big|_1^n \approx n \ln(n) - n$$

This was discovered by de Moivre

Stirling discovered the $\sqrt{2\pi n}$ factor

Euler using Euler-Maclaurin formula showed how to develop systematic corrections to this area formula

Schroeder gives another derivation on pages 390-391 that I will guide you through on homework problem.

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Outline of analyzing Einstein solid
for many ($\gg N_A$) oscillators

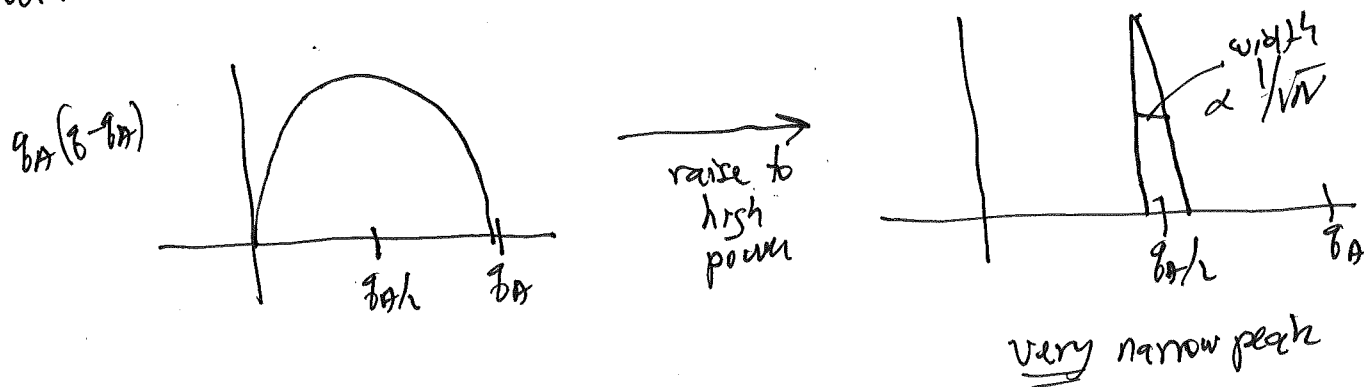
show that $\Sigma(N, \epsilon) = \binom{N-1+\epsilon}{\epsilon} \simeq \left(\frac{e\epsilon}{N}\right)^N$

when $\epsilon \gg N \gg 1$ "high temperature" limit for which a lot of energy over "few" oscillators

then $\Sigma_{\text{total}} \simeq \Sigma_A(\epsilon_A) \Sigma_B(\epsilon_B)$
 $= \Sigma(N, \epsilon_A) \Sigma(N, \epsilon_B)$
 $\simeq \left(\frac{e}{N}\right)^{2N} [\epsilon_A(\epsilon - \epsilon_A)]^N$

assume equal #
of oscillators for
simplicity
 $N_A = N_B = N$

will they show that $[\epsilon_A(\epsilon - \epsilon_A)]^N$ is extremely narrow peak
with width $1/\sqrt{N}$ for N large



High temperature approximation
to $SZ(N, g)$, $g \gg N \gg 1$

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Use Stirling and Taylor series, nice application

$$SZ(N, g) = \binom{N+g-1}{g} = \frac{(N+g-1)!}{g!(N-1)!}$$

The minus ones in $N+g-1$, $N-1$ you would guess can be dropped if $g \gg N \gg 1$ since 1 small compared to N and $N+g$

so guess that

$$SZ(N, g) = \frac{(N+g-1)!}{g!(N-1)!} \approx \frac{(N+g)!}{g! N!}$$

can justify this using our rule that "large # $\times$ very large # = very large #", For example, observe that

$$\frac{1}{(N-1)!} = \frac{N}{N!} \quad \text{and} \quad (N+g-1)! = \frac{(N+g)!}{N+g}$$

$$\text{so } \frac{(N+g-1)!}{g!(N-1)!} = \frac{(N+g)!}{g! N!} \cdot \underbrace{\frac{N}{N+g}}$$

$\approx \frac{N}{g}$, large # (or reciprocal of large #) compared to $g!$, $N!$ so ignore

good mathematical strategy: convert

(1) very large # to large # by log, simplify,
then exponentiate to recover original expression

(2) look for opportunity to identify tiny ratio $\frac{N}{g} \ll 1$
and use Taylor expansion

$$\begin{aligned}\ln S(N, g) &\approx \ln \left[\frac{(N+g)!}{N! g!} \right] = \ln[(N+g)!] - \ln N! - \ln g! \\ &\approx (N+g) \ln(N+g) - (N+g) - (N \ln N - N) - (g \ln g - g) \\ &= (N+g) \ln(N+g) - N \ln N - g \ln g\end{aligned}$$

Observe that $\ln(N+g) = \ln \left[g \left(1 + \frac{N}{g} \right) \right] = \ln g + \ln \left(1 + \frac{N}{g} \right)$
 $\approx \ln g + \frac{N}{g}$ since $N/g \ll 1$

where I used Taylor series approx $\boxed{\ln(1+x) \approx x}$ $|x| \ll 1$

So: $\ln S(N, g) = (N+g) \left[\ln g + \frac{N}{g} \right] - N \ln N - g \ln g$

$$= \left(N \ln g + g \ln g + \frac{N^2}{g} + N \right) - N \ln N - g \ln g$$

$$\approx N \left[\ln g - \ln N + 1 \right] + \underbrace{N \cdot \left(\frac{N}{g} \right)}_{\text{drop as tiny compared with } N}$$

$$\approx N \left[\ln g - \ln N + \ln(e) \right]$$

$$\approx N \ln \left[\frac{g e}{N} \right]$$

$$\Rightarrow \boxed{S = \left(\frac{g e}{N} \right)^N} \quad \text{qed}$$

$\ln(e) = 1$
by defn