

1/20/11

(1)

1st quiz next Tues

come few minutes early to class

- put everything off desk except
pen, pencil, or eraser

- no calculator

- 15 min

will post list of topics to know

Quick Review

Equilibrium macroscopic systems described by just a few macroscopic numbers, e.g. P, V, T

Criteria for equilibrium:

1. macroscopic time indep.
2. T same everywhere
3. P same everywhere unless grav, elect. fields, then P can vary
4. Fluxes are balanced when # of particles can vary
chemical potential μ
5. no-relative macroscopic motion
- ⇒ 6. properties independent of history, hardest to test
notion of internal microscopic motion scrambles
history, e.g. random collisions in gas, liquid,
random vibrations in solid

"ergodic": all possible ~~states~~ microscopic states
eventually achieved

not obvious: wood
= glass
alloys

quantum tunneling ⇒ everything liquid over long
enough times.

Review 2

For each kind of nonequilibrium, mechanisms act to restore equilibrium:

- diff in temp \Rightarrow heat conduction L^2/k
- diff in speeds \Rightarrow viscous damping L^2/ν
(relative motion)
- diff in concentrations \Rightarrow particle diffusion L^2/D
- diff in pressure \Rightarrow relative motion L^2/ν

each mech has its own independent time scale, relaxation time τ

easiest to look up dissipation coefficients

- thermal diffusivity k
 - kinematic viscosity ν
 - diffusion coeff D
- } all have units $\tau \frac{L^2}{T}$

have to wait for longest relaxation time (usually many multiples) to reach equilibrium

1/20/11 (4)

Class problem:
time to defrost Siberian mammoth
show slide

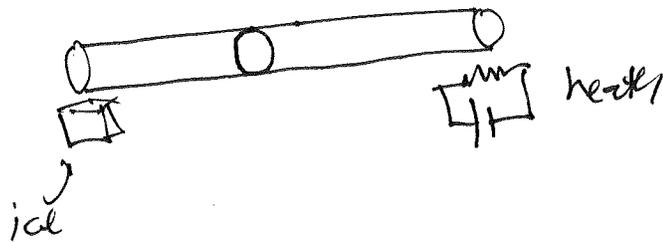
Small systems equilibrate quickly

100 nm blob of water

$$\tau_{\text{thermal}} = \frac{(10^{-7} \text{ m})^2}{2 \cdot 10^{-7} \frac{\text{m}^2}{\text{s}}} \approx 10^{-7} \text{ s} = 0.1 \mu\text{s}$$

so although most of universe non-equilibrium, small regions are in equil.

also explains why it is ok to talk about $T(t, x)$, T varies with time and space



although T strictly meaningful only for equil system, can apply to small regions of system provided

$$\Delta t > \tau$$

local region equilibrates faster than overall dynamics

1/20/11 (6)

Why L^2 Scaling?

Why Does relaxation take so long?

Simple model of random collisions gives insight

Consider single He atom in middle of box of air (N_2)

assume:

(1) travels at constant speed v until it hits another molecule

(2) assume always travels same distance d before next collision

$d = \text{mean free path}$

$\Delta t = d/v = \text{collision time}$

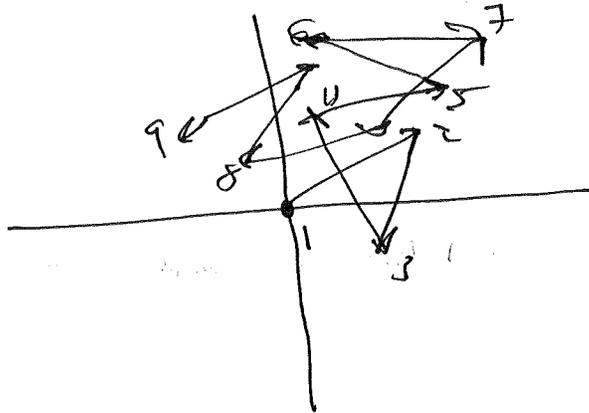
for N_2 $d \approx 380 \text{ nm} > \text{mean spacing} \approx$
 $\approx 40 \times \text{typical spacing}$

$\Delta t \approx 3 \cdot 10^{-10} \text{ s} \approx 0.3 \text{ ns}$

(3) assume each collision causes given particle to change direction randomly

Question: on average, how ~~long~~ far does particle get in time $T = N \Delta t$ $N = \# \text{ of steps}$?

Show MMA notebook
of random walk



note how plot
looks like a long
chain of links,
our theory will
apply to polymers

many repetitions suggest that, after N steps,
particle has traveled distance \sqrt{Nd} from origin
compare with Nd if no collisions, traveling in
straight line

more detailed analysis, with distribution of distance
between collisions, gives same conclusion.

spectacular example: Sun has radius of ≈ 2.2 light-seconds
light you see started off as δ rays
in core of Sun, took $\approx 100,000$ years
to diffuse to surface and escape!
core is dense, photons don't go far before
colliding with $pt, e, et \dots$

Proof that $\langle L^2 \rangle \propto N$

Concept of ensemble average

$$\vec{X} = \vec{\Delta X}_1 + \vec{\Delta X}_2 + \dots + \vec{\Delta X}_N = \sum_{i=1}^N \vec{\Delta X}_i$$

$|\vec{\Delta X}_i| = d$ all vector steps same distance

$$\begin{aligned} |\vec{X}|^2 &= \vec{X} \cdot \vec{X} = (\vec{\Delta X}_1 + \vec{\Delta X}_2 + \dots) \cdot (\vec{\Delta X}_1 + \vec{\Delta X}_2 + \dots) \\ &= \vec{\Delta X}_1^2 + \vec{\Delta X}_1 \cdot \vec{\Delta X}_2 + \vec{\Delta X}_2^2 + \dots \\ &\quad + \vec{\Delta X}_2 \cdot \vec{\Delta X}_1 \\ &= \sum_{i=1}^N \vec{\Delta X}_i^2 + 2 \sum_{\substack{i=1 \\ j=1 \\ i < j}}^N \sum \vec{\Delta X}_i \cdot \vec{\Delta X}_j \end{aligned}$$

$$|\vec{X}|^2 = Nd^2 + 2 \sum_{i=1}^N \sum_{\substack{j=1 \\ i < j}}^N \vec{\Delta X}_i \cdot \vec{\Delta X}_j$$

now carry out ensemble average: repeat N steps from origin K diff times and record

$$\begin{array}{ccc} \vec{\Delta X}_1^{(1)} & \vec{\Delta X}_1^{(2)} & \vec{\Delta X}_1^{(3)} \\ \vec{\Delta X}_2^{(1)} & \vec{\Delta X}_2^{(2)} & \vec{\Delta X}_2^{(3)} \end{array}$$

← label of experiment

Define

$$\begin{aligned} \langle F \rangle &= \text{ensemble average} \\ &= \frac{1}{K} \sum_{s=1}^K F^{(s)} \end{aligned}$$

← label tells value for k^{th} experiment starting in same way

$L^2 \propto N$ continued

Because $\langle \dots \rangle$ is an average

$$\langle a+b \rangle = \langle a \rangle + \langle b \rangle$$

$$\langle cf \rangle = c \langle f \rangle \quad c \text{ some constant.}$$

Thus

$$\langle |x|^2 \rangle = \left\langle Nd^2 + 2 \sum_{i < j} \vec{\Delta x}_i \cdot \vec{\Delta x}_j \right\rangle$$

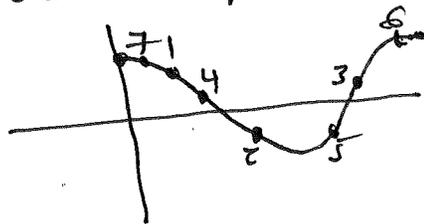
$$= \langle Nd^2 \rangle + 2 \sum_{i < j} \langle \vec{\Delta x}_i \cdot \vec{\Delta x}_j \rangle$$

$$= Nd^2 + 2 \sum_{i < j} \underbrace{\langle \cos(\theta_{ij}) \rangle}_{\text{tiny}}$$

$$\approx Nd^2 \quad \text{Q.E.D.}$$

$\frac{1}{K} \sum_{k=1}^K \cos(\theta_{ij}^{(k)})$
becomes arbitrarily
small for large K
try w/ Mathematica

because random change in direction implies $\cos(\theta_{ij})$ will
be averaged over all possible values of $0 \leq \theta_{ij} < 2\pi$,
plus and negative etc



so $X_{\text{rms}} \propto \sqrt{N} \propto \sqrt{T} \Rightarrow T \propto X_{\text{rms}}^2$

L^2 scaling

$$\langle x^2 \rangle = \left(\frac{d^2}{\Delta t} \right) T \quad \text{diffusion coefficient } D = \frac{d^2}{\Delta t}$$

Random walk is a big thing

Einstein used this idea to prove existence of atoms and measure k and N_A accurately

$$\langle x^2 \rangle = \frac{kT}{3\pi\eta a} \cdot t$$

a = radius of little sphere

η = viscosity of fluid

t = observation time

k = Boltzmann's constant

experiments give
 $R = N_A k$
 so measuring k also
 gives N_A

won Jean Perrin Physics Nobel in 1926

most cited paper of Einstein

big implications for biology: reliable processes in presence of random collisions

path is fractal: infinite length, finite area

other way to measure Avogadro's #?

Ben Franklin and tablespoon of oil